

Letter

## Preparation of a new pepper: chemoenzymatic synthesis of capsaicin oligosaccharide and 8-nordihydrocapsaicin oligosaccharide

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### Abstract

Capsaicin and 8-nordihydrocapsaicin were readily converted into the corresponding monoglucosides (capsaicin  $\beta$ -D-glucopyranoside and 8-nordihydrocapsaicin  $\beta$ -D-glucopyranoside) with tetraacetyl- $\alpha$ -D-glucose fluoride (TAGF) in the presence of  $\text{BF}_3 \cdot \text{OEt}_2$ . Furthermore, capsaicin oligosaccharides and 8-nordihydrocapsaicin oligosaccharides were synthesized from their monoglucosides by a cyclodextrin glucanotransferase (CGTase)-catalyzed glycosylation for the preparation of higher water-soluble capsaicinoids. © 2001 Elsevier Science B.V. All rights reserved.

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The hot or chili peppers are members of the genus *capsicum*. The extremely pungent oleoresin in *capsicum* is capsaicin (about 0.02% in flesh fruit and 0.5–1% in the dried ripe fruit). Capsaicin is a fat-soluble phenol. It imparts a distinctly pungent taste to water even when diluted to one part in eleven million parts of water [1]. The irritant principle of hot peppers, capsaicin, possesses extensive neurological toxicity most pronounced in the developing nervous system and focused on substance P containing neurons. It also exhibits direct skin and mucous

membrane irritant effects [2]. Furthermore, the excitatory effects of dihydrocapsaicin on nociceptive neurons in the medial thalamus were reported [3]. It was reported that, capsaicin reduced the perirenal adipose tissue weight and serum triglyceride concentration in rats by enhancing the energy metabolism through a  $\beta$ -adrenergic action [4,5]. In humans, it was reported that, the ingestion of chili sauce with meals resulted in a marked increase in energy metabolism [6]. Irrespective of such biological activities, the use of capsaicin and its derivative as food ingredients has been limited, because of its low solubility.

Glycosylation allows the conversion of water-insoluble organic compounds to the corresponding water-soluble one for improving its bio- and pharmacological

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properties [7,8]. For example, Kometani et al. demonstrated the preparation of capsaicin monoglucoside using a suspension culture of *Coffea arabica* [9].

In this paper, we report that the chemoenzymatic synthesis of the capsaicin oligosaccharides and 8-nordihydrocapsaicin oligosaccharide as new capsaicinoids which possess a higher water-solubility.

Capsaicin was purchased from Nacalai Tesque, Inc., Kyoto, Japan. 8-Nordihydrocapsaicin was purchased from Tokyo Kasei Kogyo Co. Ltd., Tokyo, Japan. Cyclodextrin glucanotransferase (CGTase from *Bacillus macerans*) was obtained from Amano Pharmaceutical Co. Ltd., Nagoya, Japan. All other reagents used were of analytical grade.  $^1\text{H}$  NMR (400 MHz) spectra were recorded on a BRUKER AMX-R400 spectrometer in methanol- $d_4$  using tetramethylsilane (TMS) as the internal standard. The FAB-MS spectrum was measured on a JEOL, the MStation JMS-700 spectrometer. The molecular weight was estimated from the  $m/z$  value of the quasimolecular ion  $[\text{M} + \text{Na}]$  peak. Tetraacetyl- $\alpha$ -D-glucose fluoride (TAGF) was prepared according to the literature procedure [10]. Capsaicin (**1**) and 8-nordihydrocapsaicin (**3**) were readily converted to the corresponding monoglucoside (**2a,4a**) with TAGF in the presence of  $\text{BF}_3 \cdot \text{OEt}_2$  [11]. As a typical run, under a nitrogen atmosphere,  $\text{BF}_3 \cdot \text{OEt}_2$  (4.0 ml) was added to a mixture of TAGF (15 mmol), substrate (**1** or **3**, 10 mmol), and 1,1,3,3-tetramethylguanidine (TMG) (17 mmol) in dry acetonitrile (15 ml, distilled from  $\text{CaH}_2$ , and stored over MS 4A) at room temperature. After the reaction mixture was stirred for 3 h, saturated aqueous  $\text{NaHCO}_3$  was added. The organic materials were extracted twice with  $\text{AcOEt}$ , washed with brine, dried over  $\text{Na}_2\text{SO}_4$ , and concentrated under reduced pressure. The viscous residue was dissolved into 20 ml of MeOH containing 400 mg of  $\text{K}_2\text{CO}_3$ . After stirring for 2 h at room temperature, the mixture was filtered and concentrated under reduced pressure. The crude residue was recrystallized twice from  $\text{H}_2\text{O}$ –MeOH to give a white powder (isolated yield: **2a**, 71%; **4a**, 78%). Selected NMR data: **2a**;  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta = 0.955$  (d, 6H), 1.36 (m, 2H), 1.62 (m, 2H), 1.98 (q, 2H), 2.18 (t, 2H), 2.32 (m, 1H), 3.30–3.48 (m, 4H), 3.66 (dd, 1H), 3.69 (dd, 1H), 3.84 (s, 3H), 4.30 (d, 2H), 4.84 (br, 1H), 5.36 (d, 1H, anomeric H:  $J = 6.7$  Hz), 5.37 (m, 2H), 6.82 (d, 1H), 6.94 (s, 1H), 7.11 (d, 1H), **4a**;  $^1\text{H}$  NMR (400 MHz,

$\text{CD}_3\text{OD}$ ):  $\delta = 0.893$  (t, 3H), 1.30 (m, 10H), 1.62 (m, 2H), 2.21 (t, 2H), 3.29–3.51 (m, 4H), 3.66 (dd, 1H), 3.69 (dd, 1H), 3.84 (s, 3H), 4.29 (d, 2H), 4.64 (br, 1H), 4.85 (d, 1H, anomeric H:  $J = 7.6$  Hz), 6.81 (dd, 2H), 6.93 (d, 1H), 7.11 (d, 1H). The FAB-MS spectrum of **2a** and **4a** gave a molecular ion at  $m/z$  490 and 478  $[\text{M} + \text{Na}]$ , respectively. A large coupling constant ( $J = 6.6$  and 7.6 Hz) of the anomeric proton in **2a** and **4a** suggested the  $\beta$  configuration for the anomeric center. Therefore, **2a** and **4a** could be identified as capsaicin monoglucoside (capsaicin  $\beta$ -D-glucopyraoside) and 8-nordihydrocapsaicin monoglucoside (8-nordihydrocapsaicin  $\beta$ -D-glucopyraoside), respectively (Fig. 1).

As the synthetic monoglucosides (**2a** and **4a**) prepared had a resemble water-solubility to those of **1** and **3** (data not shown), the further glucosylation to the sugar moiety in **2a** and **4a** was necessary to increase the solubility of their capsaisinoids. For the further glucosylation of the monoglucosides, we planed to use a cyclodextrin glucanotransferase (CGTase) which catalyzed the transfer reaction of sugar from “donor sugar” to “acceptor sugar” to form oligosaccharides [12,13].

Soluble starch (40 g) and  $\text{CaCl}_2$  (20 mg) were added to 100 ml of distilled water (pH 6.0), and heated until the starch was completely dissolved (it became a viscous solution). The substrate (50 mg, **2a** or **4a**) and CGTase (1800 U) were added to the viscous starch solution. After stirring for 24 h at 55°C, the mixture was extracted twice with *n*-butanol, and concentrated under reduced pressure to give a white powder (Fig. 2).

The crude products were separated and purified using an HPLC equipped with a Crestpak C18S (JASCO,  $\phi$  4.6 mm  $\times$  150 mm, UV 280 nm, flow rate: 1.0 ml/min, mobile phase:  $\text{CH}_3\text{CN}:\text{H}_2\text{O} = 13:7$ , column temperature: 40°C). When we used  $\alpha$ -,  $\beta$ - and  $\gamma$ -cyclodextrins as sugar donors, the further glucosylation for the monoglucosides (**2a** and **4a**) were not proceeded effectively by the CGTase-catalyzed reaction. The optimum temperature of the reaction was 55°C. The CGTase-catalyzed glucosylation toward **1** and **3** was not observed.

Five products were produced from **4a** as shown in Fig. 3. The FAB-MS analyses of isolated **4b–f** indicated the formation of 8-nordihydrocapsaicin oligosaccharide (**4b**; diglucoside, **4c**; triglucoside, **4d**;

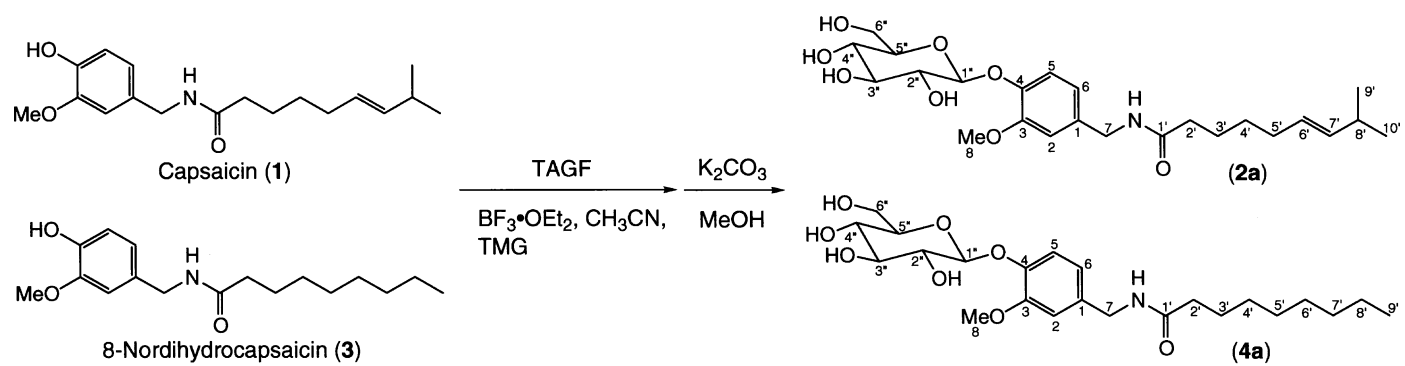


Fig. 1. Synthesis of capsaicin β-D-glucopyranoside (**2a**) and 8-nordihydrocapsaicin β-D-glucopyranoside (**4a**) from **1** or **3**.

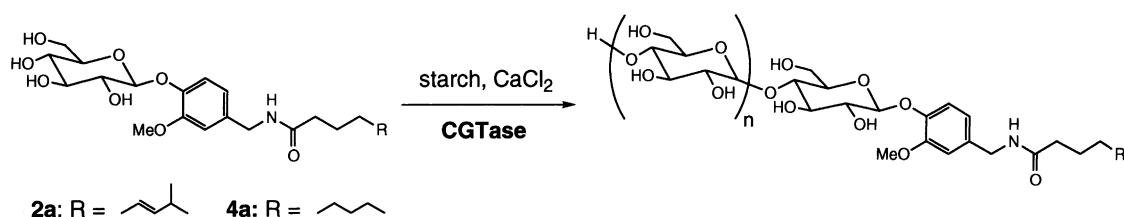


Fig. 2. Enzymatic synthesis of capsaisin oligosaccharides and 8-nordihydrocapsaicin oligosaccharides from the corresponding monoglucosides (**2a** and **4a**).

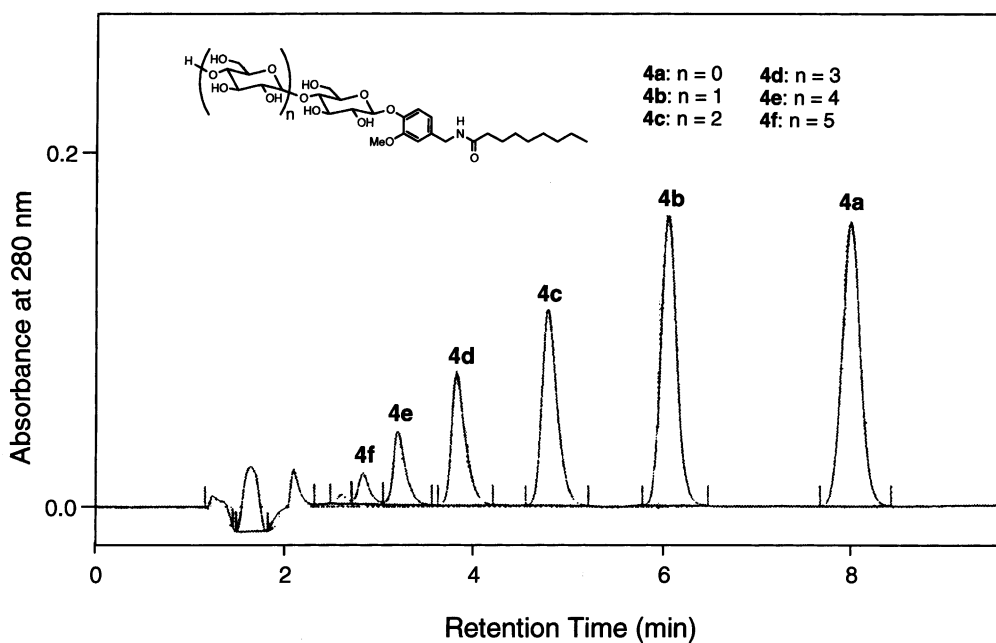


Fig. 3. The HPLC chromatogram of the products from **4a** by CGTase-catalyzed glycosylation.

tetraglucoside, **4e**; pentaglucoside, and **4f**; hexaglucoside), because the  $m/z$  values were 478, 640, 802, 964, 1126, and 1288, respectively. The isolated yields of products **4a**, **4b**, **4c**, **4d**, **4e**, and **4f** are 13, 12, 8, 6, 4, and 3 mg, respectively (from 50 mg of **4a**). In the same manner, four oligosaccharides (**2b**; diglucoside, **2c**; triglucoside, **2d**; tetraglucoside, **2e**; pentaglucoside) were produced from **2a** (chromatographic data not shown). The isolated yields of products **2a**, **2b**, **2c**, **2d**, and **2e** are 16, 13, 10, 4, and 2 mg, respectively (from 50 mg of **2a**).

Thus, the enzymatic synthesis of new capsaicin and 8-nordihydrocapsaicin oligosaccharides was achieved

by the CGTase-catalyzed glycosylation in the presence of starch as a sugar donor.

The application of this chemoenzymatic system to a large-scale (gram scale) production of the capsaicin oligosaccharides and their derivatives is currently under investigation and will be reported in a future paper.

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